

Elastic fluid and rotating fluid

Boyle's law:- If the tempⁿ remains constant, the pressure p varies inversely as the volume v , i.e.

$$p \propto \frac{1}{v}$$

$$\text{OR } p = \frac{\lambda}{v}, \lambda \text{ is a constant}$$

$$\text{OR } pv = \lambda \rightarrow \textcircled{1}$$

Charles's law:- If the pressure remains constant, the volume increases by a definite fraction of the volume at 0°C for every degree centigrade through which the tempⁿ increases.

Thus if v_0 and v be the volumes of the gas at 0°C and $t^\circ\text{C}$, we have

$$v = v_0 (1 + \alpha t) \rightarrow \textcircled{1}$$

where $\alpha = \frac{1}{273}$ is called the coefficient of expansion and t is temperature.

Some results for gases:- (Elastic fluid)

If tempⁿ remains constant, then for a gas

$$pv = \lambda \text{ (Boyle's law)}$$

Also, if ρ be the density and P the pressure, we have the mass

$$m = v\rho \text{ (constant)}$$

$$\text{Now, } p = \frac{\lambda}{v} = \frac{\lambda}{m} \rho$$

$$\therefore p = k\rho \rightarrow \textcircled{1} \text{ where } \frac{\lambda}{m} = k \text{ is a constant.}$$

If the tempⁿ varies then by Charles's law,

$$v = v_0 (1 + \alpha t), \text{ where } \alpha \text{ is the coefficient of}$$

expansion ($\alpha = \frac{1}{273}$), t is the tempⁿ in centigrade and v_0 is the volume at $t = 0^\circ\text{C}$.

Now, the mass remains unchanged

$$\text{i.e. } \rho v = \rho_0 v_0$$

$$\therefore \frac{\rho_0}{\rho} = \frac{v}{v_0} = \frac{v_0 (1 + \alpha t)}{v_0} = (1 + \alpha t)$$

$$\text{or } p_0 = \rho (1 + \alpha t) \rightarrow (3)$$

Again, if p and p_0 be the pressure and density at 0°C ,

$$\text{then } p = k p_0 \rightarrow (4)$$

And if ρ be the density when the temp is $t^\circ\text{C}$, where p remains unchanged, then

$$p_0 = \rho (1 + \alpha t) \rightarrow (5)$$

Hence from (4), we get

$$p = k \rho (1 + \alpha t) \\ = k \rho \alpha \left(t + \frac{1}{\alpha}\right)$$

$$\therefore p = k \rho T, \text{ where } T = \left(t + \frac{1}{\alpha}\right) \text{ and } k \alpha = k.$$

T is called the absolute temp.

Elastic fluid :- ($p = k \rho$) when temp is constant.

(06) Determination of pressure :-

* 2015 A mass of elastic fluid is at rest under the action of given forces. To determine the pressure at any pt.

2018 AS question :- A mass of elastic fluid is at rest under the action of given forces. Determine the pressure at any pt. for both the cases when the temp remains constant and when it varies.

Let the components of forces along the direction of x -, y -, z -axes be X , Y , Z respectively per unit mass. Since the fluid is at rest under the action of the given forces, therefore the pressure eqn at any pt. is given by

$$dp = \rho (X dx + Y dy + Z dz) \rightarrow (1)$$

Case 1 :- When the temp remains constant, then

$$p = k \rho \rightarrow (2) \quad \text{For elastic fluid}$$

where k is a constant and ρ the density.

Dividing (1) by (2), we get

$$\frac{dp}{p} = \frac{X dx + Y dy + Z dz}{k}$$

Integrating, we get

$$\log p = \frac{1}{k} \int (x dx + Y dy + z dz) + \log c$$

$$\Rightarrow \log \frac{p}{c} = \frac{1}{k} \int (x dx + Y dy + z dz)$$

$$\Rightarrow \frac{p}{c} = e^{\frac{1}{k} \int (x dx + Y dy + z dz)}$$

$$\Rightarrow p = c e^{\frac{1}{k} \int (x dx + Y dy + z dz)} \rightarrow (3)$$

Now, if the forces are conservative, then
 $x dx + Y dy + z dz = -dv$, where v is the
potential function.

$$\therefore (3) \Rightarrow p = c e^{\frac{1}{k} \int (-dv)} = c e^{-k/v}$$

which determines the pressure at any pt.

Case 2:- When the tempⁿ varies, then

$$p = k \rho T \rightarrow (4) \text{ (when tempⁿ is not constant)}$$

where k is a constant and T the absolute tempⁿ.

Now, dividing (3) by (4), we get

$$\frac{dp}{p} = \frac{x dx + Y dy + z dz}{kT}$$

Integrating, we get

$$\log p = \frac{1}{k} \int \frac{x dx + Y dy + z dz}{T} + \log c_1 \rightarrow (5)$$

If the forces are conservative, then

$$x dx + Y dy + z dz = -dv$$

$$\therefore (5) \Rightarrow \log \frac{p}{c_1} = \frac{1}{k} \int \frac{(-dv)}{T}$$

$$\Rightarrow \frac{p}{c_1} = e^{-\frac{1}{k} \int \frac{dv}{T}}$$

$$\text{or } p = c_1 e^{-\frac{1}{k} \int \frac{dv}{T}}$$

which gives the pressure at any point.

Condition of Equilibrium:- of an elastic fluid.

The necessary and sufficient condition for
an elastic fluid may maintain equilibrium
which must be satisfied under a given

system of forces x, γ, z acting per unit mass of the fluid and || to the rectangular co-ordinate axes is

$$x \left(\frac{\partial \gamma}{\partial z} - \frac{\partial z}{\partial y} \right) + \gamma \left(\frac{\partial z}{\partial x} - \frac{\partial x}{\partial z} \right) + z \left(\frac{\partial x}{\partial y} - \frac{\partial \gamma}{\partial x} \right) = 0$$

(8.8, 90, 92) M (statement only)

Rotating fluid:-

Eqⁿ of pressure at a point:-

Let the liquid revolve with a uniform angular velocity ω about the vertical axis, the rotation being such that there is no relative displacement of its particles, i.e. the liquid revolves like a rigid body about the axis.

Let us take the axis of rotation as the axis of z . Now we consider a particle of mass m at a distance r from the axis. This particle is moving in a circle of radius r with uniform angular velocity ω , so that the tangential acceleration due to rotation vanishes and the normal acceleration, i.e. $r\omega^2 (= r\dot{\omega}^2)$ acts along the inward drawn ~~non~~ normal (i.e. towards the axis being $m\omega^2 r$).

If we apply a force $m\omega^2 r$ away from the axis the force due to rotation is balanced and the system is in static equilibrium.

The components of this applied force $m\omega^2 r$ along the axes are

$$m\omega^2 r \cos \theta = m\omega^2 r \cdot \frac{x}{r} = m\omega^2 x, \text{ along } x\text{-axis.}$$

$$m\omega^2 r \sin \theta = m\omega^2 r \cdot \frac{y}{r} = m\omega^2 y, \text{ along } y\text{-axis.}$$

If x, γ, z be the external forces per unit mass acting on the fluid, then

$$\text{Total external force along } x\text{-axis} = m(x + \omega^2 x)$$

$$\text{Total external force along } \gamma\text{-axis} = m(\gamma + \omega^2 y)$$

$$\text{and total external force along } z\text{-axis} = mz.$$

\therefore The pressure p at a point (x, y, z) is given

∴ The differential eqn for pressure is

$$dp = \rho [\omega^2 x dx + \omega^2 y dy - g dz]$$

Integrating, we get

$$p = \rho \left[\frac{1}{2} \omega^2 (x^2 + y^2) - g z \right] + A \rightarrow \textcircled{1}$$

This gives pressure at any point.

Now, putting

$p = \text{constant} = c$ (say), the surfaces of equal

pressure are of the form

$$c = \rho \left[\frac{1}{2} \omega^2 (x^2 + y^2) - g z \right] + A$$

$$\frac{1}{2} \omega^2 (x^2 + y^2) - g z = \frac{c - A}{\rho} = c_1 \text{ (say)}$$

$$\text{OR } x^2 + y^2 = \frac{2}{\omega^2} (g z + c_1) = \frac{2g}{\omega^2} \left(z + \frac{c_1}{g} \right)$$

$$\text{OR } x^2 + y^2 = \frac{2g}{\omega^2} (z - c_2) \text{ where } \frac{c_1}{g} = -c_2$$

which clearly represents paraboloids of revolution

of latus rectum $\frac{2g}{\omega^2}$.

(08)M

2017

Ex) An open vessel containing liquid is made to revolve about a vertical axis with uniform angular velocity. Find the form of the vessel and its dimensions that it may be just emptied.

Soln:- Let the liquid revolve with uniform angular velocity ω then pressure at any pt. is given by,

$$dp = \rho[(\omega^2 x dx + \omega^2 y dy) - g dz] \rightarrow (1)$$

[Since, the force along x-axis is $m\omega^2 x$ and the force along y-axis is $m\omega^2 y$ and along z-axis, the force is only gravity which is $-g$ (positive z-axis taken upward).] Integrating (1), we get

$$p = \rho \left[\frac{1}{2} \omega^2 (x^2 + y^2) - g z \right] + c \rightarrow (2)$$

Let us take the lowest point of the vessel to be the origin. Then since the vessel is just emptied, there is no liquid or liquid pressure at the lowest point.

$$\therefore p = 0 \text{ when } x = y = z = 0$$

$$\text{i.e. } c = 0$$

Hence (2) becomes

$$p = \rho \left[\frac{1}{2} \omega^2 (x^2 + y^2) - g z \right]$$

Again since the vessel is just emptied, the inner surface of the vessel coincides with the free surface of the liquid. So putting

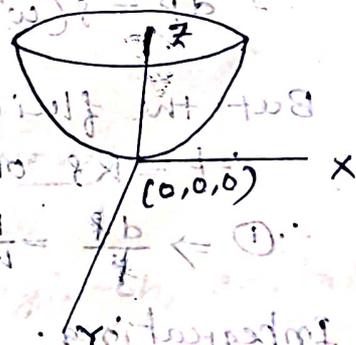
$p = 0$, the free surface of the liquid, i.e. the

inner surface of the vessel is given by

$$\frac{1}{2} \omega^2 (x^2 + y^2) - g z = 0$$

$$\Rightarrow x^2 + y^2 = \frac{2g}{\omega^2} z$$

which is a paraboloid of revolution of latus rectum $\frac{2g}{\omega^2}$.



$$\begin{aligned} \therefore M &= \int_{-a}^a \int_{-a}^a \int_{-a}^a \rho \, dx \, dy \, dz \\ &= 8 \rho_0 \int_0^a \int_0^a \int_0^a e^{-[(\mu - \omega^v)(x^v + y^v) + \mu z^v] / 2k} \, dx \, dy \, dz \\ &\quad \because \int_{-a}^a f(x) \, dx = 2 \int_0^a f(x) \, dx \text{ if } f(-x) = f(x) \\ &= 8 \rho_0 \int_0^a e^{-(\mu - \omega^v) \frac{x^v}{2k}} \, dx \times \int_0^a e^{-(\mu - \omega^v) \frac{y^v}{2k}} \, dy \end{aligned}$$

$$\begin{aligned} &\times \int_0^a e^{-\frac{\mu z^v}{2k}} \, dz \\ &= 8 \rho_0 \cdot \frac{\sqrt{\pi}}{2 \sqrt{\frac{\mu - \omega^v}{2k}}} \cdot \frac{\sqrt{\pi}}{2 \sqrt{\frac{\mu - \omega^v}{2k}}} \cdot \frac{\sqrt{\pi}}{2 \sqrt{\frac{\mu}{2k}}} \end{aligned}$$

$$\int_0^a e^{-ax^v} \, dx = \frac{\sqrt{\pi}}{2 \sqrt{a}}$$

$$\text{OR } M = \rho_0 \frac{(2k\pi)^{3/2}}{(\mu - \omega^v) \sqrt{\mu}} \rightarrow \textcircled{3}$$

Now, from $\textcircled{2}$

$$2k \log \frac{\rho_0}{\rho} = (\mu - \omega^v)(x^v + y^v) + \mu z^v$$

$$\text{OR } 2k \log \left\{ \frac{M(\mu - \omega^v) \sqrt{\mu}}{\rho (2k\pi)^{3/2}} \right\} = (\mu - \omega^v)(x^v + y^v) + \mu z^v$$

$$\text{OR } k \log \left\{ \frac{M^v (\mu - \omega^v)^v \mu}{\rho^v (2k\pi)^3} \right\} = \mu (x^v + y^v + z^v) - \omega^v (x^v + y^v)$$

$$\begin{aligned} \text{OR } k \log \left\{ \frac{\mu (\mu - \omega^v)^v}{8\pi^3} \cdot \frac{M^v}{\rho^v k^3} \right\} &= \mu (x^v + y^v + z^v) - \omega^v (x^v + y^v) \\ &= \mu (x^v + y^v + z^v) - \omega^v (x^v + y^v) \quad \text{PROVED} \end{aligned}$$

1.2.2.2.2.2

2015
10) Ex)

A mass M of a gas at uniform temperature is distributed through all space, and at rest each point (x, y, z) the components of force per unit mass are $-Ax, -By, -cZ$.

The pressure and density at the origin are p_0 and ρ_0 resp. $\therefore p, \rho \propto \rho_0 M^{\frac{1}{3}} = 8\pi^3 \rho_0^{\frac{1}{3}}$.

Soln

The pressure eqn is

$$dp = \rho \cdot (x dx + y dy + z dz)$$

$$= -\rho (Ax dx + By dy + cZ dz) \quad \text{--- (1)}$$

Since the temperature is uniform.

So by the gas law, $p = k\rho$

$$\Rightarrow dp = k d\rho \rightarrow \text{(2)}$$

from (1) and (2) we get,

$$k d\rho = -\rho (Ax dx + By dy + cZ dz)$$

$$\Rightarrow \frac{d\rho}{\rho} = -\frac{1}{k} (Ax dx + By dy + cZ dz)$$

Integrating we get,

$$\log \rho = -\frac{1}{2k} (Ax^2 + By^2 + cZ^2) + \log C_1$$

$$\Rightarrow \rho = C_1 e^{-\frac{1}{2k} (Ax^2 + By^2 + cZ^2)} \quad \text{--- (3)}$$

but at the origin $x=y=z=0$ then $\rho = \rho_0$.

$$\therefore (3) \Rightarrow \rho_0 = C_1$$

$$\text{again, (3)} \Rightarrow \rho = \rho_0 e^{-\frac{1}{2k} (Ax^2 + By^2 + cZ^2)}$$

Considering the volume element $\delta x, \delta y, \delta z$ at $P(x, y, z)$

\therefore mass of gas acting the volume element $= \rho \delta x \delta y \delta z$

$$\therefore M = \int_{x=-a}^a \int_{y=-a}^a \int_{z=-a}^a \rho dx dy dz$$

$$= \int_{x=-a}^a \int_{y=-a}^a \int_{z=-a}^a \rho_0 e^{-\frac{1}{2k} (Ax^2 + By^2 + cZ^2)} dx dy dz$$

$$= 8\rho_0 \int_0^a e^{-(Ax^2/2k)} dx \int_0^a e^{-(By^2/2k)} dy \int_0^a e^{-(cZ^2/2k)} dz$$

$$= 8\rho_0 I_1 I_2 I_3$$

(1) - T P U U

Now $I_1 = \int_0^{\infty} e^{-(Ax^2/2k)} dx$

Let $\frac{Ax^2}{2k} = t$

$\Rightarrow \frac{2Ax dx}{2k} = dt$

$\Rightarrow dx = \frac{k dt}{Ax}$

$= \int_0^{\infty} e^{-t} \frac{k dt}{A \sqrt{2kt}}$

$= \sqrt{\frac{k}{2A}} \int_0^{\infty} e^{-t} t^{-1/2} dt$

$= \sqrt{\frac{k}{2A}} \int_0^{\infty} e^{-t} t^{\frac{1}{2}-1} dt$

$= \sqrt{\frac{k}{2A}} \sqrt{\frac{\pi}{2}}$

$= \sqrt{\frac{k\pi}{2A}}$

ii) $I_2 = \sqrt{\frac{k\pi}{2B}}$

$I_3 = \sqrt{\frac{k\pi}{2C}}$

$\therefore M = 8 \rho_0 \sqrt{\frac{k\pi}{2A}} \left(\sqrt{\frac{k\pi}{2B}} + \sqrt{\frac{k\pi}{2C}} \right)$

$= 8 \rho_0 \sqrt{\frac{k^3 \pi^3}{8ABC}}$

$M^3 = 8 \rho_0^3 \frac{k^3 \pi^3}{ABC}$ — (4)

Again from the gas law

$p = k \rho$

but at the origin $p = p_0$ and $\rho = \rho_0$

$\therefore p_0 = k \rho_0$

$\Rightarrow k = \frac{p_0}{\rho_0}$

(4) $\Rightarrow M^3 = 8 \rho_0^3 \frac{p_0^3 \pi^3}{\rho_0^3 ABC}$

$\Rightarrow ABC \rho_0 M^3 = 8 \pi^3 p_0^3$

Proved

2019 • **Example 45.** A thin sphere of radius a just filled with water rotates about a vertical diameter with angular velocity $\omega = \sqrt{\left(\frac{2g}{3a}\right)}$; prove that the pressure at any point of the surface of equal pressure which cuts the sphere at right angles is $\frac{3}{4}\rho ga$, ρ being the density of water.

Sol. The differential equation of the pressure at any point is

$$dp = \rho [\omega^2 (x dx + y dy) - g dz].$$

Integrating,

$$p = \rho \left[\frac{1}{2} \omega^2 (x^2 + y^2) - gz \right] + c$$

The sphere is just filled with water, therefore at the highest point when

$$x = 0, y = 0, z = a, p = 0; \therefore c = \rho ga$$

Thus

$$p = \rho \left[\frac{1}{2} \omega^2 (x^2 + y^2) - gz + ag \right].$$

This gives pressure at any point.

Now the surface of equal pressure, where pressure = $\frac{3}{4}\rho ga$ is given by

$$\frac{3}{4}\rho ga = \rho \left[\frac{1}{2} \cdot \frac{2g}{3a} (x^2 + y^2) - gz + ag \right]$$

or
$$x^2 + y^2 - 3gz + \frac{3}{4}a^2 = 0,$$

which is clearly a surface cutting the sphere

$$x^2 + y^2 + z^2 = a^2$$

at right angles.

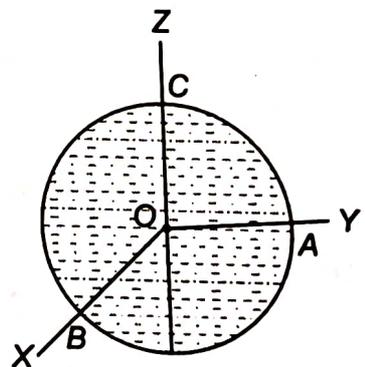


Fig. 192